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### **Chapter 4 Assessment Chemistry Answer Key**

AP Chemistry Chapter 4 Answers - Zumdahl 485 Apply the rules in Table 42 a  $\text{KMnO}_4$  is composed of  $\text{K}^+$  and  $\text{MnO}_4^-$  ions Assign oxygen a value of -2, which gives manganese a +7 oxidation state since the sum of oxidation states for all atoms in  $\text{MnO}_4^-$  must equal the -1 charge on  $\text{MnO}_4^-$ .  $\text{K}^+$ , +1; O, Chemistry Assignment Sheet Chapters 4 and 5

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## **Chemistry Chapter 4 Test Answers - Reliefwatch**

AP Chemistry Chapter 4 Answers - Zumdahl 450 The solubility rules referenced in the following answers are from Table 41 of the text The phrase "slightly soluble" is interpreted to mean insoluble and the phrase "marginally soluble" is interpreted to mean soluble a Soluble (Rule 3) b Soluble (Rule 1) c Insoluble (Rule 4) d

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