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Holt Chemistry Chapter 3. a chemical compound always contains the.... mass cannot be created or destroyed in.... when two elements combine to form two o.... subatomic particle with a negative char.... law of definite proportion. law of conservation of mass. law of multiple proportions. electron.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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Holt McDougal Modern Chemistry Chapter 3 Test. Honors Chemistry Chapter 3 Test. STUDY. PLAY. law of conservation of mass. mass is neither created nor destroyed during ordinary chemical reactions or physical changes $HH+O = HH O$. law of definite proportions.

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State the following measured quantities in the units indicated. a. 5.2 cm of magnesium ribbon in millimeters b. 0.049 kg of sulfur in grams c. 1.60 mL of ethanol in microliters d. 0.0025 g of vitamin A in micrograms e. 0.020 kg of tin in milligrams f. 3 kL of saline solution in liters

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Name CHAPTER 3 TEST continued Date Class FILL IN THE BLANK
Write the correct term (or terms) in the space provided. 9, If a particular compound is composed of elements A and B, the ratio of the mass of B to the mass of A will always be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

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continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. Modern Chemistry Chapter 3 Test Answer Key

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