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following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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chromium ion must be followed by a Roman numeral indicating its charge.

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Stock system name such as iron(III)
sulfate, the Roman numeral tells us (a)
how many atoms of Fe are in one

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