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# Physical Science Atomic Structure Answers Sheet

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## **Physical Science Atomic Structure Answers**

Atomic Structure. The atom consists of three components: Protons, Neutrons, and Electrons. Atomic Number. The order of an element in Mendeleev's table of the elements; Often represented by the symbol  $Z$  and it dictates the number of protons in the nucleus. Atomic Symbol.

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Q. Pure forms of these elements are stored in oil so they won't react with oxygen and water in the air.

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Atomic Structure Objective: To understand how protons, neutrons, and electrons are arranged in atoms and how the atom's mass, charge, and identity is affected by the number of these particles.

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## **Physical Science Atomic Structure Answers Sheet**

In this topic we learn about the structure of an atom. We cover proton, neutron, and electron properties and their locations within an atom. We also cover setting up Bohr models for elements, atomic number, atomic mass, and valence electrons as well. Watch the video and answer the questions in your guided notes taker. Do great on the topic quiz!

### **a. Atomic Structure - FSIcourses**

answer choices Electrons vibrate in fixed locations around the nucleus. Electrons travel around the nucleus in fixed energy

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levels with energies that vary from level to level. Electrons travel around the nucleus in fixed energy levels with equal amounts of energy.

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Chapter 4 Physical Science Atomic Structure Vocab. Nucleus. Proton. Neutron. Electron. the dense, positively charged mass located in the center of an.... A positively charged subatomic particle that is found in the n.... A neutral subatomic particle that is found in the nucleus of a....

## **test physical science chapter 4 atomic structure ...**

Atomic Structure. Atom V Element ATOMIC THEORY Democritus (4th century B.C) said the universe was made of invisible units called atoms In 1808, an Englishman named John Dalton proposed a theory: 1. every element is made of tiny unique

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particles called atoms that cannot be subdivided 2. atoms of the same element have the same mass & atoms of different elements have different masses 3. compounds contain atoms of more than one element 4. in any particular compound, atoms of different elements ...

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grade 10 physical sciences. grade 10 revision questions and answers; atomic structure; energy; instantaneous speed velocity & equations of motion; introduction to chemistry; isotope and relative atomic mass; quantitative aspects of chemical reaction; skills and facts; giant molecular substances; matter and material; giant metallic and giant ...

## **ATOMIC STRUCTURE - Physical Sciences Break 1.0**

Georgia Milestones - Physical Science EOC: Atomic Structure



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Two models of atomic structure are in use today: the Bohr model and the quantum mechanical model. The quantum mechanical model is based on mathematics. Although it is more difficult to understand than the Bohr model, it can be used to explain observations made on complex atoms. A model is useful because it helps you [...]

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